

VIBRANT ACADEMY

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VSAT (VIBRANT SCHOLARSHIP ADMISSION TEST)

OFFLINE SAMPLE PAPER

XII PASSED (GOOGOL COURSE-2025-26)

Time : 2 Hours

Maximum Marks : 240

Please read the instructions carefully.

INSTRUCTIONS

A. General :

- **1.** This booklet is your Question Paper containing **60** questions.
- 2. The **question paper CODE** is printed on the right hand top corner of this sheet and on the back page of this booklet.
- **3.** The question paper contains **2** blank pages for your rough work. No additional sheets will be provided for rough work.
- **4.** Blank papers, clipboards, log tables, slide rules, calculators, cellular phones, pagers and electronic gadgets in any form are not allowed to be carried inside the examination hall.
- Fill in the boxes provided below on this page and also write your Name and Regn. No. in the space provided on the back page of this booklet.
- 6. The answer sheet, a machine-readable Objective Response Sheet (**ORS**), is provided separately.
- 7. DO NOT TAMPER WITH / MUTILATE THE ORS OR THE BOOKLET.
- 8. Do not open the question-paper booklet before being instructed to do so by the invigilators.
- B. Question paper format
- C. Marking scheme

Read the instructions printed on the back page of this booklet.

Name of the candidate
UID Number

I have read all the instructions and shall abide by them.
I have verified all the information filled in by the candidate.

Signature of the Candidate
Signature of the invigilator

<u>USEFUL DATA</u>

Atomic weights: AI = 27, Mg = 24, Cu = 63.5, Mn = 55, CI = 35.5, O = 16, H = 1, P = 31, Ag = 108, N = 14, Li = 7, I = 127, Cr = 52, K=39, S = 32, Na = 23, C = 12, Br = 80, Fe = 56, Ca = 40, Zn = 65.5, Ti = 48, Ba = 137, U = 238, Co = 59, B = 11, F = 19, He = 4, Ne = 20, Ar = 40, Mo = 96, g = 10 m/s²

PART I : PHYSICS SECTION - I Single Correct Choice Type

	This section contains 14 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4) for its answer, out of which ONLY ONE is correct.						
1.	The correct set of diamagnetic substances among the following sets are						
	(A) Aluminium, sodium calcium and oxygen	(B) Bismuth, copper, lead and silicon					
	(C) Cobalt, nickel, gadolinium and aluminium	(D) Silver, niobium, magnesium and calcium					

- 2. Two springs have force constant k_A and k_B such that $k_B = 2k_A$. The four ends of the springs are stretched by the same force. If energy stored in spring A is E, then energy stored in spring B is :
 - (A) $\frac{E}{2}$ (B) 2E (C) E (D) 4E
- 3. Two identical cylinders have a hole of radius a (a<<R) at its bottom. A ball of radius R is kept on the hole and water is filled in the cylinder such that there is no water leakage from bottom. In case-1 water is filled upto height h and in second case it is filled upto height 2h. If F_1 is force by liquid on sphere in case-1 and F_2 is force by liquid on sphere in case-2 then.



4. A plano convex lens (refractive indices μ_1) fits exactly into a plano concave lens (refractive indices μ_2). Their plane surfaces are parallel to each other. R is the radius of curvature of the curved surface of the lenses, then focal length of the combination is

(A)
$$\frac{R}{\mu_1 - \mu_2}$$
 (B) $\frac{2R}{\mu_2 - \mu_1}$ (C) $\frac{2R}{2(\mu_1 - \mu_2)}$ (D) $\frac{R}{2 - (\mu_1 + \mu_2)}$

5. In each of the three arrangements, the block of mass m₁ is being pulled left with constant velocity. There is no friction any where. The strings are light and inextensible and pulleys are massless. The ratio of the speed of the block of mass m₂ in the three cases respectively is :



(A)	2	:	1	:	4	
(C)	4	:	2	:	1	

(B) 2 : 4 : 1(D) Can not be calculated

6. An open capillary tube is lowered in a vessel with mercury. The difference between the levels of the mercury in the vessel and in the capillary tube $\Delta h = 4.6$ mm. What is the radius of curvature of the mercury meniscus in the capillary tube? Surface tension of mercury is 0.46 N/m, density of mercury is 13.6 gm/cc.

(A) $\frac{1}{340}$ m (B) $\frac{1}{680}$ m (C) $\frac{1}{1020}$ m (D) Information insufficient

7. A block of mass m is attached to two spring of spring constant k_1 and k_2 as shown in figure. The block is displaced by x towards right and released. The velocity of the block when it is at its mean position .



8. The system of two weights with masses m₁ and m₂ are connected with weightless spring as shown. The system is resting on the support S. The support S is quickly removed. The accelerations of each of the weights right after the support S is removed are.



(A) $a_1 = 0, a_2 = \frac{(m_1 + m_2)g}{m_2}$	(B) $a_1 = 0, a_2 = \frac{(m_1 + m_2)g}{m_1}$
(C) $a_1 = \frac{(m_1 + m_2)g}{m_1}$, $a_2 = 0$	(D) a ₁ = 0, a ₂ = 0

9. The following figure shows two situations in which a uniform round rigid body is released from rest from the positions shown, such that it is just able to loop the loop without leaving contact with the track. Assuming that radius of the track is large in comparison to the radius of round body, the ratio h_1 / h_2 .



- (A) must be greater than 1
- (B) must be less than 1
- (C) must be equal to 1
- (D) can be greater than or less than 1, depending on the moment of inertia of the round body.
- **10.** A conducting soap bubble having a radius a, thickness t(<< a) is charged to a potential V. If now, the bubble collapses and forms a droplet, then the potential of the droplet is

(A)
$$V_{\text{droplet}} = V\left[\frac{a}{3t}\right]^{\frac{1}{3}}$$
 (B) $V_{\text{droplet}} = V\left[\frac{a}{t}\right]^{\frac{1}{3}}$ (C) $V_{\text{droplet}} = V\left[\frac{3a}{t}\right]^{\frac{1}{3}}$ (D) $V_{\text{droplet}} = V\left[\frac{3a}{4t}\right]^{\frac{1}{3}}$

11. The energy of a photon is E whih is equal to the kinetic energy of a proton. If λ_1 be the de-Broglie wavelength of the proton and λ_2 be the wavelength of the photon, then photon, then the ratio $\frac{\lambda_1}{\lambda_2}$ is proportional to (A) E⁰ (B) E^{1/2} (C) E⁻¹ (D) E⁻²



- **12.** Which of the following set of characteristics of a material is suitable for using it as a dielectric in a capacitor?
 - (A) Large dielectric constant and high dielectric strength
 - (B) Large dielectric constant and low dielectric strength
 - (C) Small dielectric constant and high dielectric strength
 - (D) Small dielectric constant and low dielectric strength
- **13.** A monoatomic gas of molar mass M₀ is filled into an insulated container which is moving along a horizontal surface with velocity a v. If the container is stopped suddenly, then the change in temperature of the gas is

(A)
$$\frac{M_0^2 v^2}{2R}$$
 (B) $\frac{M_0^2 v^2}{3R}$ (C) $\frac{2M_0^2 v^2}{3R}$ (D) $\frac{3M_0^2 v^2}{2R}$

14. A conducting ring of radius r and resistance R is placed in a region of uniform time varying magnetic field B which is perpendicular to the plane of the ring. If the magnetic field is changing at a rate α, then the current induced in the ring is

(A)
$$\frac{\pi r^2 \alpha}{2R}$$
 (B) $\frac{2\pi r \alpha}{R}$ (C) $\frac{\pi r \alpha}{R}$ (D) $\frac{\pi r^2 \alpha}{R}$

SECTION - II Numerical Answer Type

This section contains **SIX (06)** questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than **two decimal** places, **truncate/round-off** the value to **TWO** decimal places. The answer to each of the questions is a **Two decimal** integer, ranging from 00.00 to 99.99. The appropriate bubbles below the respective question numbers in the ORS have to be darkened. For example, if the correct answers to question numbers X, Y, Z and W (say) are 11, 7, 11.25 and 1.2, respectively, then the correct darkening of bubbles will look like the following. **Way 11.00, 07.00, 11.25, 01.20**

X	Y	Z	W
0 0 • 0 0 1 • 1 1 1 2 2 • 2 2 3 3 • 3 3 4 4 • 4 4 5 5 • 5 6 6 6 • 6 6 7 7 7 7 8 8 • 8 8 9 9 • 9 9	0 · 0 0 1 1 · 1 1 2 2 · 2 2 3 3 · 3 3 4 4 · 4 4 5 5 · 5 5 6 6 · 6 6 7 7 · 7 7 8 8 · 8 8 9 9 · 9 9 9	0 0 • 0 0 1 1 1 1 2 2 • 2 2 3 3 • 3 3 4 4 • 4 4 5 5 • 5 5 6 6 • 6 6 7 7 • 7 7 8 8 • 8 8 9 9 • 9 9 9	0 0 • 0 0 1 1 • 1 1 2 2 • 2 2 3 3 • 3 3 4 4 • 4 4 5 5 • 5 5 6 6 • 6 6 7 7 • 7 7 8 8 • 8 8 9

- **15.** Two simple pendulums of length 5 m and 20 m are given small displacements in the same direction at the same time. Find the minimum number of oscillations, the shorter pendulum has completed, when the phase difference between them becomes zero again
- **16.** Two particles are projected vertically upwards from the surface of the earth with velocities $v_1 = \sqrt{\frac{2gR}{3}}$ and

 $v_2 = \sqrt{\frac{4gR}{3}}$ respectively. If the maximum heights attained by the two particles are h₁ and h₂ respectively,

then calculate the ratio $\frac{h_1}{h_2}$.

17. In the above arrangement, each side of the cube have the same resistance and it is known that the effective resistance between A and B is $\frac{5}{9}\Omega$. Now, if the resistor between A and B is removed, then find the new effective resistance (in Ω) between the same two points.



18. What is the percentage error in the physical quantity A if it is related to four other physical quantities a, b, c and d as A = $\frac{a^2b^3}{d\sqrt{c}}$. The percentage error in measurement of a, b, c and d are 1%, 3%, 2% and 1% respectively.

- **19.** An ac source of 50 V (r.m.s value) is connected across a series R C circuit. If the r.m.s voltage across the resistor is 40V, find then the r.m.s voltage across the capacitor (in volt)
- **20.** A particle of mass 0.01 kg travels along a space curve with velocity given by $(4\hat{i} + 16\hat{k})ms^{-1}$. After sometime, its velocity becoems $(8\hat{i} + 20\hat{j})ms^{-1}$ due to the action of a conservative force. Then find the work done on the particle during this interval of time (in Joule)

PART II : CHEMISTRY SECTION - I

	This section cont answer, out of wh	Single Co ains 14 multiple choice qu iich ONLY ONE is correct	uestions. Each question t.	has 4 choices (1), (2), (3) and (4) for its		
21.	Which of the follo (A) FeCl ₂ and Cut (C) VOCl ₂ and Fe	wing pair has same value Cl ₂ Cl ₂	of magnetic moment (sp (B) VOCl ₂ and Cu (D) FeCl ₂ and Mn	agnetic moment (spin only)? (B) VOCl ₂ and CuCl ₂ (D) FeCl ₂ and MnCl ₂ $(\Delta_f G^\circ)_{ZnO} = -320 \text{ kJ mol}^{-1}$			
22.	For the silver-zind Zn(s) + Ag ₂ O (s) - Given : $(\Delta_f G^\circ)$ Ag What is the E° of	button cell, net reaction → ZnO (s) + 2Ag(s) $_2$ O = -20 kJ mol ⁻¹ ; button cell?	is $(\Delta_{\rm f}{ m G}^\circ)_{ m ZnO}$ = -320				
	(A)3.10 V	(B) 1.55 V	(C) 1.1 V	(D) 0.775 V			
23.		$\begin{array}{c} O \\ CI \\ \underline{AICI_3} \\ \Delta \end{array} \\ Produce$	t (P)				

Major product 'P' should be



- 24. Which of the following statement is incorrect regarding refining of copper by electrolysis?
 - (A) $\rm CuSO_4$ 15% solution and 5% $\rm H_2SO_4$ solution are used as electrolyte.
 - (B) Impure 'Cu' is used as cathode and Pure 'Cu' is used as Anode.
 - (C) More electropositive impurities like Fe, Zn, Ni etc dissolved in electrolyte solution.
 - (D) Less electropositive metals such as Ag, Au and Pt are collected below anode called anode mud.

Total number of hydrogen atoms in one litre of water is :									
(A) 55.55 × N _A	(B) 55.55 × 2 × N _A								
(C) 55.55	(D) 55.55 × 3 × N _A								
Which of the following compound gives positive lodoform Test?									
(A) 3-Hexanone	(B) 2-Hexanone	(C) Hexanal	(D) 1-Butanol						
Which of the following is more stable ?									
(A) $[Fe(H_2O)_6]^{3+}$	(B) [FeF ₆] ^{3–}	(C) $[Fe(C_2O_4)_3]^{3-}$	(D) [FeCl ₆] ³⁻						
	Total number of hydrog (A) 55.55 × N _A (C) 55.55 Which of the following (A) 3-Hexanone Which of the following (A) $[Fe(H_2O)_6]^{3+}$	Total number of hydrogen atoms in one litre of(A) $55.55 \times N_A$ (B) $55.55 \times 2 \times N_A$ (C) 55.55 (D) $55.55 \times 3 \times N_A$ Which of the following compound gives positive(A) 3-Hexanone(B) 2-HexanoneWhich of the following is more stable ?(A) $[Fe(H_2O)_6]^{3+}$ (B) $[FeF_6]^{3-}$	Total number of hydrogen atoms in one litre of water is :(A) $55.55 \times N_A$ (B) $55.55 \times 2 \times N_A$ (C) 55.55 (D) $55.55 \times 3 \times N_A$ Which of the following compound gives positive lodoform Test ?(A) 3-Hexanone(B) 2-Hexanone(C) HexanalWhich of the following is more stable ?(A) [Fe(H_2O)_6]^{3+}(B) [FeF_6]^{3-}(C) [Fe(C_2O_4)_3]^{3-}						

28.
$$(P)$$

$$(P)$$

$$(P)$$

$$(H_3)$$

Product 'P' should be



29. What is the largest wavenumber in the Pfund series of He⁺ ion?

(A) $\frac{25}{R}$	(B) 4 R	(C) $\frac{4R}{25}$	(D) None of these
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30. Which of the following oxy acid of phophorous is a reducing agent and monobasic? (A) H_3PO_2 (B) H_3PO_3 (C) H_3PO_4 (D) $H_4P_2O_6$

- **32.** Which of the compound react with $NaHCO_3$ to give CO_2



33. Which compound is having more heat of combustion



- Resistance of a conductivity cell filled with a solution of an electrolyte of concentration 0.1 M is 100 ohm. This conductivity of this solution is 1.29 Sm⁻¹. Resistance of the same cell filled with 0.02 M of the same solution if the electrolyte is 520 ohm. The molar conductivity of 0.02 M solution of electrolyte would be.
 (A) 124 × 10⁻⁴ Sm² mol⁻¹
 (B) 1240 × 10⁻⁴ Sm² mol⁻¹
 - (C) 1.24×10^{-4} Sm² mol⁻¹ (D) 12.4×10^{-4} Sm² mol⁻¹

SECTION - II

Numerical Answer Type

This section contains **SIX (06)** questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than **two decimal** places, **truncate/round-off** the value to **TWO** decimal places. The answer to each of the questions is a **Two decimal** integer, ranging from 00.00 to 99.99. The appropriate bubbles below the respective question numbers in the ORS have to be darkened. For example, if the correct answers to question numbers X, Y, Z and W (say) are 11, 7, 11.25 and 1.2, respectively, then the correct darkening of bubbles will look like the following. **Way 11.00, 07.00, 11.25, 01.20**



- **35.** A system absorbed 20 kJ of heat and work done on the surrounding is 10 kJ. The change in internal energy is equal to (in kJ).
- 36. Find the number of different type of Functional groups



- **37.** One mol of non-ideal gas undergoes a change of state (1 atm, 4 L, 100 K) to (3 atm, 6 L, 275 K) with a change in internal energy (ΔU) = 25 L-atm. The change in enthalpy (ΔH) of the process in L-atm.
- 38. Find out numerical value of expression [x ÷ y] for elements : O, F, Br, S and C. where , x = Number of element(s) having less electron gain enthalpy than chlorine y = Number of element(s) having higher electronegativity than chlorine.

- **39.** Calculate pH of an aqueous solution of 0.02 M, 2L of $Ba(OH)_2 + 0.02$ M, 3L of NaOH + 0.02 M, 5L of HCl at 90°C temperature (Given : K_w at 90°C = 10⁻¹²)
- **40.** pOH of 10^{-1} M CH₃COOH (aq) solution will be (given : K_a = 10^{-5})

PART III : MATHEMATICS SECTION - I Single Correct Choice Type

	This section contains 14 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4) for its answer, out of which ONLY ONE is correct.								
41.	If G(x) = $-\sqrt{25 - x^2}$ then $\lim_{x \to 1} \left(\frac{G(x) - G(1)}{x - 1} \right) =:$								
	(A) $\frac{1}{\sqrt{24}}$	(B) $\frac{1}{5}$	(C) _{- \sqrt{24}}	(D) None of these					
42.	Given A = $\sin^2\theta$ + cos	${}^{4}\!\theta$ then for all values of θ							
	(A) $1 \le A \le 2$	$(B) \ \frac{3}{4} \le A \le 1$	(C) $1 \le A \le \frac{13}{6}$	$(D) \ \frac{3}{4} \le A \le \frac{13}{6}$					
43.	The scalar product of	$\vec{A}.((\vec{B}+\vec{C})\times(\vec{A}+\vec{B}+\vec{C}))$	equal to :						
	(A) 0	(B) [ĀBC] + [BCA]	(C) [ĀBC]	(D) –[Ā Ē Ĉ]					
44.	The area of the triang $\left(1,\sqrt{3}\right)$ is :	gle formed by the positive	x-axis and the normal ar	ad the tangent to the $x^2 + y^2 = 4$ at					
	(A) $\sqrt{3}$	(B) 2√3	(C) 3 √3	(D) 4 √ 3					

45.The function $f(x) = 2\ell n |x| - x |x|$ is increasing on the interval.
(A) (0, 1)(B) $(0, \infty)$ (C) (-1, 1)(D) (-1, 0)

46. Let g(x) be a differentiable function on R and $\int_{sint}^{1} x^2 g(x) dx = (1 - sint)$, where $t \in \left(0, \frac{\pi}{2}\right)$. Then the value of $g\left(\frac{1}{\sqrt{2}}\right)$ equals.

(A)
$$\frac{1}{2}$$
 (B) $\frac{1}{\sqrt{2}}$ (C) 2 (D) 1

47. The shortest distance between line y - x = 1 and curve $x = y^2$, is :

(A)
$$\frac{3\sqrt{2}}{8}$$
 (B) $\frac{8}{3\sqrt{2}}$ (C) $\frac{4}{\sqrt{3}}$ (D) $\frac{\sqrt{3}}{4}$

48. Let $A = \begin{bmatrix} \beta & -1 \\ 1 & 2\beta \end{bmatrix}$ and det.(A^4) = 16, then the product of all possible real values of β equals.

(A) $\frac{1}{2}$	(B) $\frac{-1}{2}$	(C) 0	(D) 2
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49. The value of the parameter 'a' such that the area bounded by $y = a^2x^2 + ax + 1$, coordinate axes and the line x = 1 attains its least value, is equal to

(A)
$$-\frac{1}{4}$$
 (B) $-\frac{1}{2}$ (C) $-\frac{3}{4}$ (D) -1

50. If $f(x) = \sin x + \cos x$ and $g(x) = x^2 - 1$, then g(f(x)) is invertible in the domain

(A)
$$\left[0,\frac{\pi}{2}\right]$$
 (B) $\left[-\frac{\pi}{4},\frac{\pi}{4}\right]$ (C) $\left[-\frac{\pi}{2},\frac{\pi}{2}\right]$ (D) $[0,\pi]$

51.
$$\int e^{x} \left(\log_{e}(3x+2) + \frac{9}{(3x+2)^{2}} \right) dx \text{ is :}$$
(A) $e^{x} \left(\log_{e}(3x+2) - \frac{3}{3x+2} \right) + C$
(B) $e^{x} (\log_{e}(3x+2)) + C$
(C) $e^{x} \left(\log_{e}(3x+2) + \frac{1}{3x+2} \right) + C$
(D) $e^{x} \left(\log_{e}(3x+2) + \frac{3}{3x+2} \right) + C$

52. The value of of
$$\int_{0}^{\frac{\pi}{2}} x \left| \sin^2 x - \frac{1}{2} \right| dx$$
 is equal to

(A)
$$\frac{\pi}{4}$$
 (B) $\frac{\pi}{8}$ (C) $\frac{\pi}{2}$ (D) $\frac{3\pi}{4}$

53. If m is a non-zero real number then $\int \frac{x^{5m-1} + 2x^{4m-1}}{(x^{2m} + x^m + 1)^3} dx$, is equal to [Note : C denotes constant of integration]

(A)
$$\frac{x^{5m} - x^{4m}}{2m(x^{2m} + x^m + 1)} + C$$

(B) $\frac{2m(x^{5m} + x^{4m})}{(x^{2m} + x^m + 1)^2} + C$
(C) $\frac{x^{4m}}{2m(x^{2m} + x^m + 1)^2} + C$
(D) $\frac{x^{5m}}{2m(x^{2m} + x^m + 1)^2} + C$

54. If the function $f(x) = \begin{cases} -x, & x < 1 \\ a + \cos^{-1}(x+b), & 1 \le x \le 2 \end{cases}$ is differentiable at x = 1, then $\frac{a}{b}$ is equal to

(A)
$$\frac{\pi - 2}{2}$$
 (B) $\frac{\pi + 2}{2}$ (C) $\frac{-\pi - 2}{2}$ (D) $-1 - \cos^{-1}(2)$

SECTION - II Numerical Answer Type

This section contains **SIX (06)** questions. The answer to each question is a **NUMERICAL VALUE**. If the numerical value has more than **two decimal** places, **truncate/round-off** the value to **TWO** decimal places. The answer to each of the questions is a **Two decimal** integer, ranging from 00.00 to 99.99. The appropriate bubbles below the respective question numbers in the ORS have to be darkened. For example, if the correct answers to question numbers X, Y, Z and W (say) are 11, 7, 11.25 and 1.2, respectively, then the correct darkening of bubbles will look like the following. **Way 11.00, 07.00, 11.25, 01.20**

X	Y	Z	W
0 0 • 0 0 1 1 • 1 1 2 2 • 2 2 3 3 • 3 3 4 4 • 4 4 5 5 • 5 5 6 6 • 6 6 7 7 7 7 7 8 8 8 8 8 9 9 • 8 8	 0 1 1<	0 0 0 0 1 1 1 1 2 2 2 2 2 3 3 3 3 3 4 4 4 4 5 5 5 5 6 6 6 6 7 7 7 7 8 8 6 8 9 9 6 9	0 · 0 0 1 1 · 1 1 2 2 · 2 2 3 3 · 3 3 4 4 · 4 4 5 5 · 5 5 6 6 · 6 6 7 7 · 7 7 8 8 · 8 8 9 9 9 · 9 9 9 9

55. The value of $tan\left(\sum_{r=1}^{\infty} tan^{-1}\left(\frac{4}{4r^2+3}\right)\right)$ is equal to

56. If
$$\alpha$$
 and β are roots of the equation $x^2 - 7x + 1 = 0$, then value of $\frac{1}{(\alpha - 7)^2} + \frac{1}{(\beta - 7)^2}$ is

- 57. If $\lim_{x \to \infty} \left(1 + \frac{a}{x} \frac{4}{x^2} \right)^{2x} = e^3$, then 'a' is equal to : [Note : e denotes Napier's constant]
- **58.** Let a, b and c be any real numbers, Suppose that there are real numbers x, y and z not all zero such that x = cy + bz, y = az + cx and z = bx + ay, then $a^2 + b^2 + c^2 + 2abc$ is equal to

59. If the sum
$$\frac{3}{1^2} + \frac{5}{1^2 + 2^2} + \frac{7}{1^2 + 2^2 + 3^2} + \dots +$$
 up to 20 terms is equal to $\frac{k}{21}$, then k is equal to

60. Locus of the point of intersection of the pair of perpendicular tangents to the circles $x^2 + y^2 = 1$ and $x^2 + y^2 = 7$ is the director circle of the circle with radius.

Name of the candidate	UID Number							

Β. **Question paper format :** 9. The question paper consists of 3 parts (Physics, Chemistry and Mathematics). Each part has 2 sections. 10. Section I contains 14 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D) for its answer, out of which only one is correct. 11. Section II contains 06 questions. The answer to each of the questions is a Number upto Two Decimal. The answer will have to be appropriately bubbled in the ORS as per the instructions given at the beginning of the section. C. Marking Scheme : 12. For each question in Section I, you will be awarded 4 marks if you darken the bubble corresponding to the correct answer and zero mark if no bubble is darkened. In case of bubbling of incorrect answer, minus one (-1) mark will be awarded. 13. For each question in Section-II, you will be awarded 4 marks if you darken the bubble corresponding to the correct answer and zero mark if no bubble is darkened. In case of bubbling of incorrect answer, minus one (-1) mark will be awarded.

					A	NSWE	ER KE	(
					PA	RT I :	PHYSI	CS					
1.	В	2.	А	3.	В	4.	А	5.	А	6.	В	7.	D
8.	А	9.	В	10.	А	11.	В	12.	А	13.	В	14.	D
15.	02.00	16.	00.25	17.	01.25	18.	13.00	19.	30.00	20.	00.96		
	PART II : CHEMISTRY												
21.	В	22.	В	23.	А	24.	В	25.	В	26.	В	27.	С
28.	А	29.	С	30.	А	31.	А	32.	С	33.	D	34.	А
35.	10.00	36.	07.00	37.	39.00	38.	02.50	39.	09.60	40.	11.00		
					PART	III : MA	THEM	ATICS					
41.	А	42.	В	43.	А	44.	В	45.	А	46.	С	47.	А
48.	В	49.	С	50.	В	51.	А	52.	В	53.	С	54.	В
55.	02.00	56.	47.00	57.	01.50	58.	01.00	59.	120.00	60.	02.00		

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